

Synthesis and Spectral Characterization of Novel 2,3-Disubstituted Quinazolin-4(3H) one Derivatives

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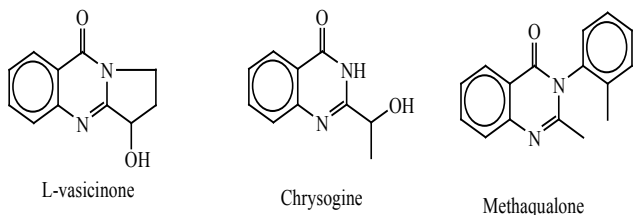
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Abstract New series of 2,3-disubstituted quinazolin-4(3H)-one were synthesized via the reaction of the readily obtainable 2-thioxo-3-phenyl-quinazolin-4(3H)-one 1 with ethyl chloroacetate followed by hydrazinolysis to afford the hydrazide 3 which allowed to react with different electrophilic reagents such as carbon disulphide, phenyl isothiocyanate, β -diketones, anhydrides, acrylonitrile, ethyl cinnamate, dimethyl acetylene dicarboxylate, aldehydes, arylidene malononitrile and lauroyl chloride. Some of the newly synthesized compounds showed promising anti-inflammatory activity.

Keywords Pyrrolizyl-, 1,3,4-Oxadiazolyl Quinazolinone, Anti-Inflammatory, Michael Reaction

1. Introduction

Several Quinazolinone derivatives were synthesized as potential antimicrobial[1,2], anticancer[3-7], anti-inflammatory[8-11], and antimalarial[12] agents. Quinazolin-4(3H)-one is a frequently encountered unit in natural products such as L-vasicinone[13], chrysoeine[14,15] and drugs as Methaqualone[16], i.e. molecules based on quinazolinone and quinazolinone exhibit a multitude of interesting pharmacological activities[17-19].



Aforementioned findings promoted the authors to synthesize a varieties of 2, 3-disubstituted quinazolinone derivatives via the utility of the key starting material 2-thioxo-3-phenyl quinazolin 4(3H)-one[20,21].

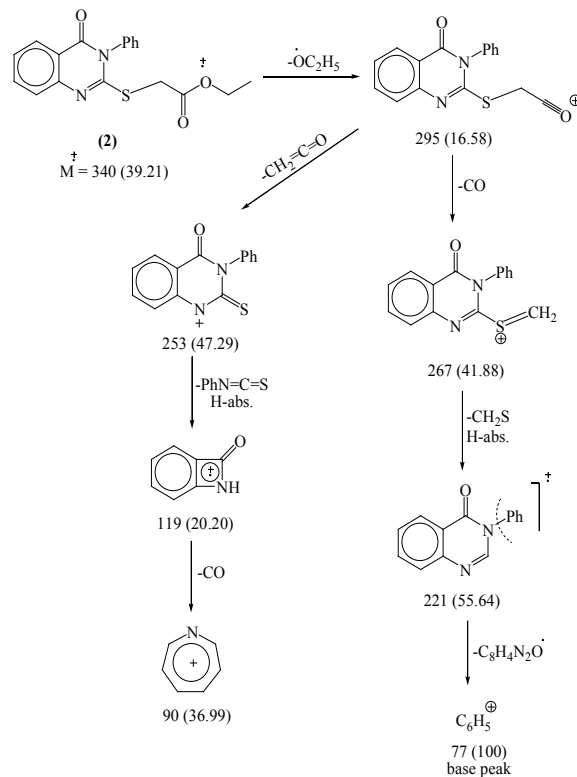
2. Results and Discussion

In this investigation, Ethyl-2-(4-oxo-3-phenyl-3, 4-dihydroquinazolin-2-yl thio)acetate 2 was obtained in fairly good yield upon treatment of quinazolin-2-thione derivative 1 with ethyl chloroacetate, in the presence of fused so-

dium acetate in refluxing ethanol. (Scheme 1)

The structure of compound 2 was confirmed from the study of its spectral data (c.f. Exp.). The highest recorded peak at $m/z = 340$ (34.21%) represent the molecular ion peak.

In previous work[22], the reaction of ethyl S-(heteryl) thioglycollate with hydrazine in refluxing ethanol involves the elimination of this group and substituted by the hydrazine group.



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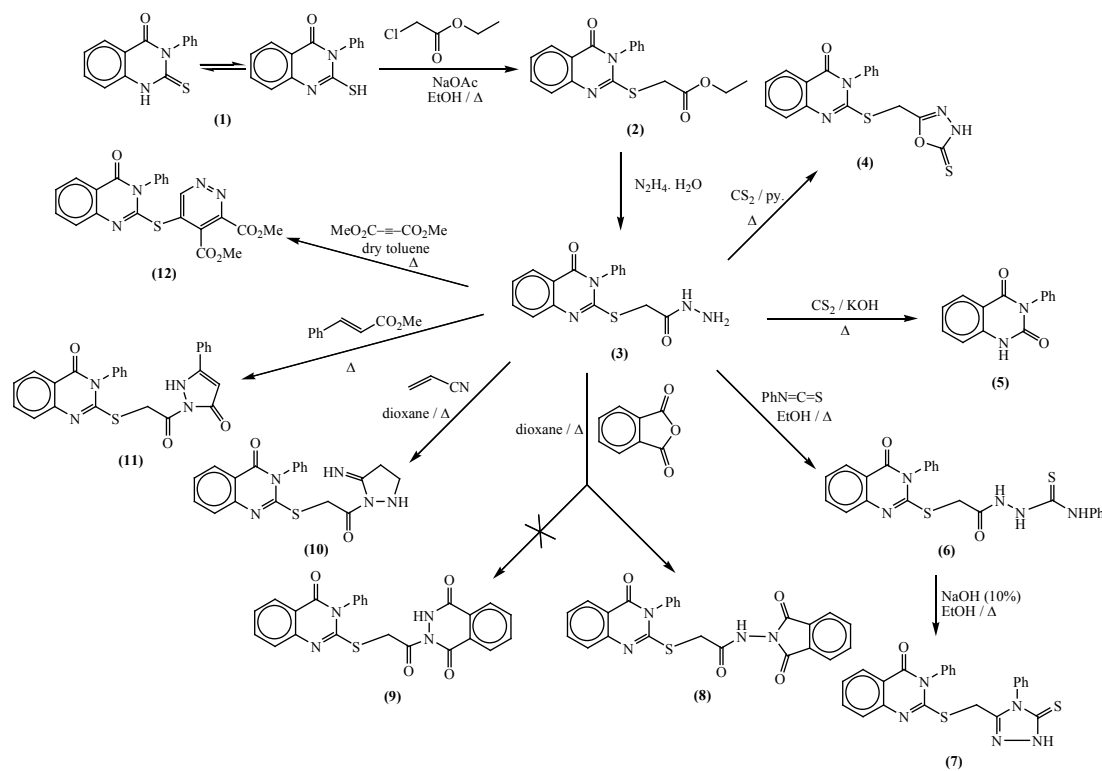
When compound 2 was allowed to react with hydrazine hydrate afforded the

2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-ylthio) aceto-hydrazide 3 which obtained via the nucleophilic nitrogen attack of the hydrazine moiety to the carbonyl group of the ester group through tetrahedral mechanism.

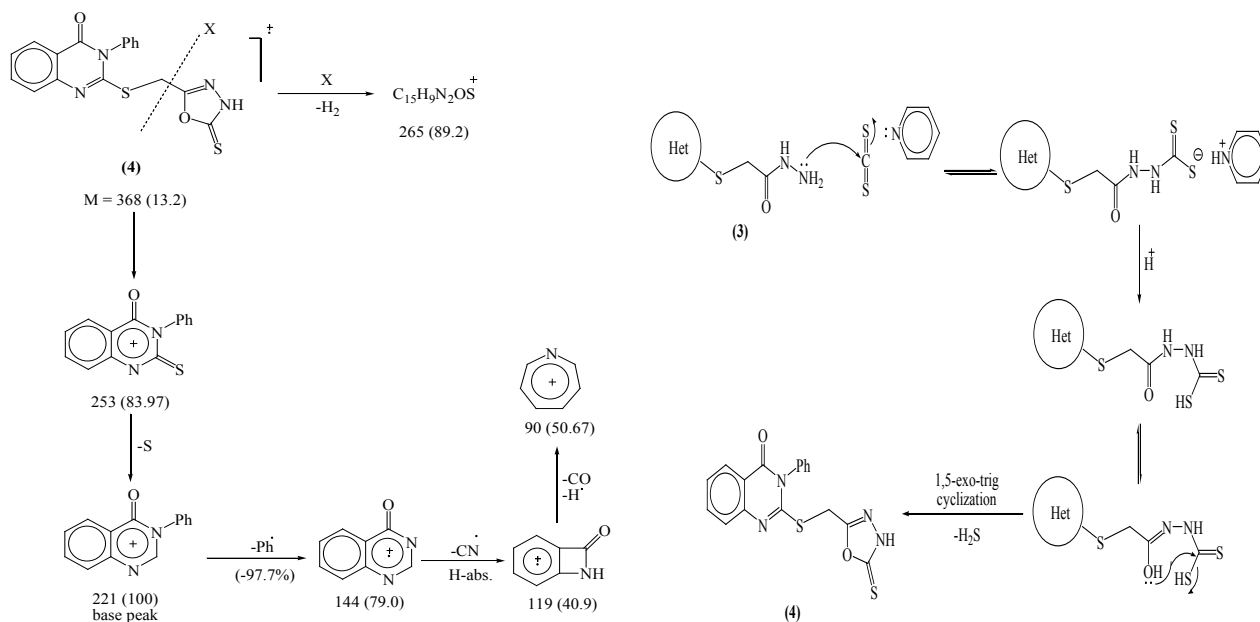
The structure of 3 was established through spectroscopic (IR, ^1H NMR and MS) beside the correct elemental analysis. (c.f. Exp)

3-Phenyl-2-[(5-thioxo-4,5-dihydro-1,3,4-oxadiazol-2-yl)

methyl thio]quinazolin-4(3H)one 4 was obtained in fairly good yield upon refluxing the hydrazide 3 with carbon disulphide in pyridine on water bath for 8hrs. The structure 4 was deduced from the correct analytical and spectroscopic data (IR, ^1H NMR). Full analysis for the mass spectrum of 4 shows the correct molecular ion peak at $m/z = 368$ (13.2%). The reaction is smoothly proceeded through tetrahedral pathway followed by 1,5-exo-trig cyclization to give 4. (Scheme 2)

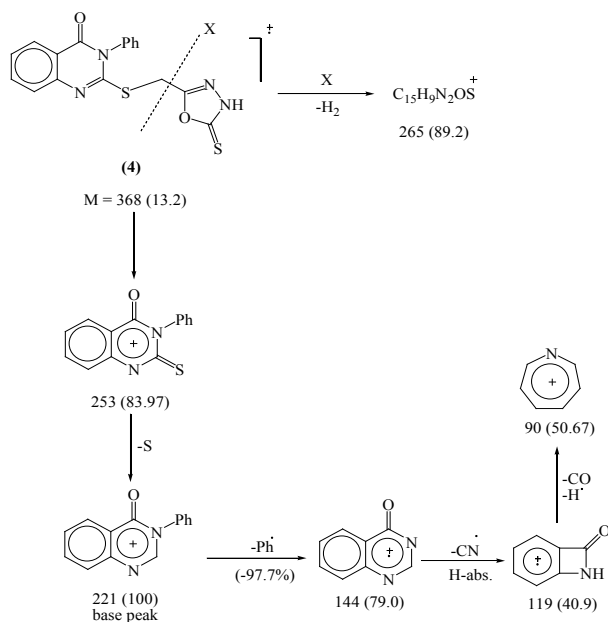


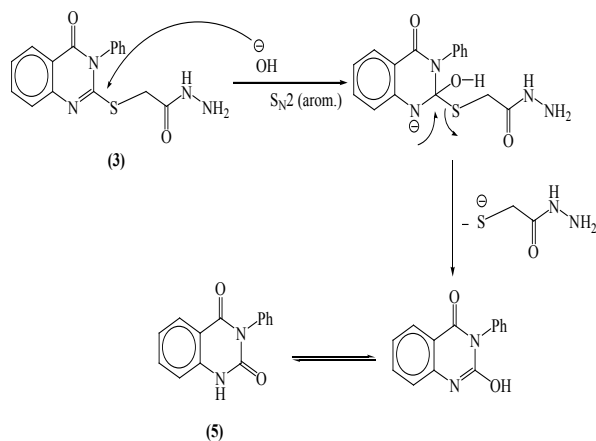
Scheme 1



Scheme 2

The mass fragmentation of 1,3,4-oxadiazolthione derivative 4





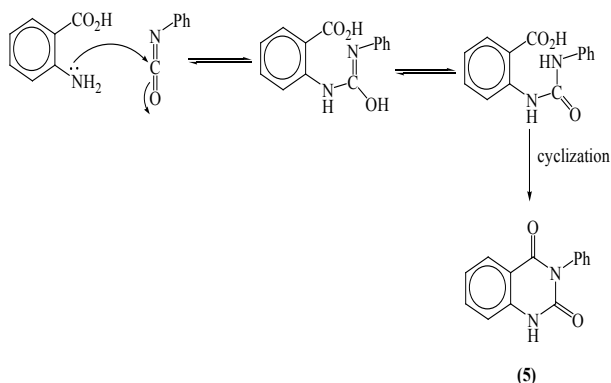
Scheme 3

In contrast, when the hydrazide 3 was treated with carbon disulphide in ethanolic potassium hydroxide, no combination was detected between the hydrazide 3 and the carbon disulphide and the solid separated was identified as the sulfur free compound which assigned as 3-phenyl-quinazolin-2,4(1H,3H)-dione 5. This is may be due to the nucleophilic substitution reaction by the nucleophilic hydroxyl group at the C₂ position of the quinazolinone and the thiomethyl hydrazide group was knocked out via S_N2 aromatic mechanism. (Scheme 3)

The structure 5 was confirmed from the study of the IR and mass spectrum together with chemical evidence and rigidly confirmed by m.p. comparison with that reported[23].

The EI fragmentation of the mass spectrum of compound 5 shows the radical cation peak at m/z = 238 (100%) which represent the molecular ion and the base peak. (c.f. Exp.)

Furthermore, compelling evidence for the structure 5 is forthcoming from the preparation of authentic sample by the reaction of anthranilic acid with phenyl isocyanate in refluxing pyridine (Scheme 4).



Scheme 4

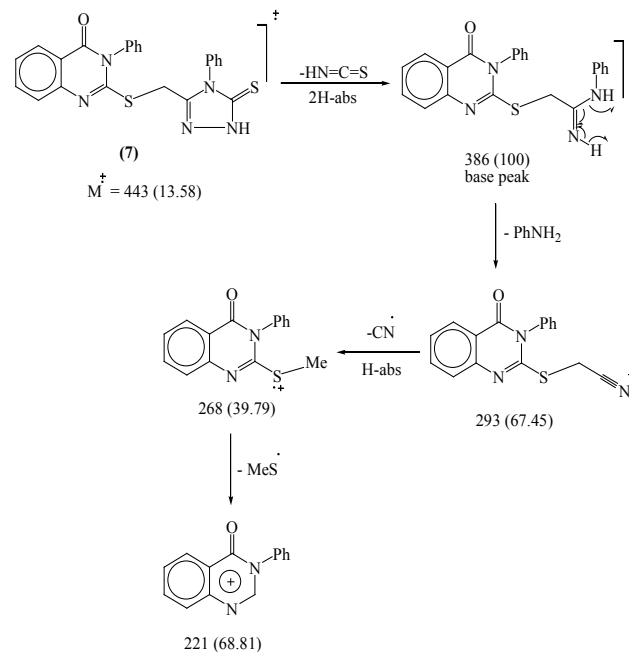
Phenylisothiocyanate was added to the hydrazide 3 in boiling ethanol and yielded 1-[2-(4-oxo-3-phenyl-3, 4-dihydroquinazolin-2-yl thio)acetyl]-4-phenyl thiosemicarbazide 6.

The structure of 6 was confirmed from the correct analytical and spectroscopic data. The highest recorded peak in

the mass spectrum of 6 at m/z = 461 (21.73%) represented the molecular ion peak which upon loss of N-phenyl thiosemicarbazide molecule afforded the base peak at m/z = 294 (100%). (c.f. Exp.)

4-Phenyl-3-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl methyl thio)-1,2,4-triazol-5(1H)thione 7 was achieved through the reaction of compound 6 with alcoholic sodium hydroxide. (Scheme 1) The structure 7 was confirmed from the correct analytical and spectroscopic data. The EI-MS revealed the molecular ion peak at m/z = 443 (13.58%).

When the hydrazide 3 was allowed to react with phthalic anhydride in refluxing dioxane yielded N-(1,3-dioxoisindolin-2-yl)-2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio) acetamide 8.



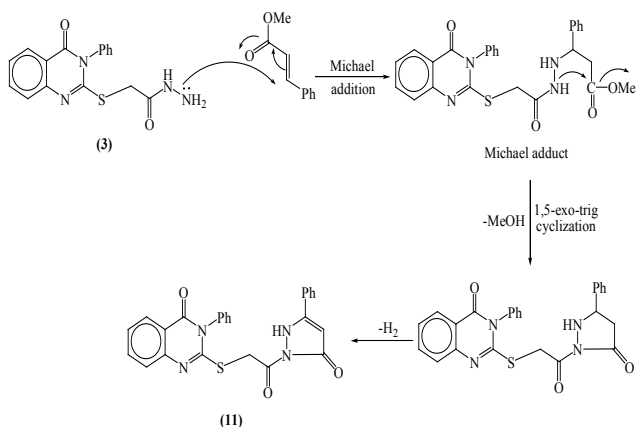
EI-MS of compound 7

The structure of 8 was confirmed from the correct analytical and spectroscopic data. No evidence for the formation of the phthalazine derivative 9 was detected, since, the IR spectrum of the product shows the carbonyl vibrational coupling bands at 1793 and 1738cm⁻¹ together with ν_{C=O} (quinazolinone) at 1684 cm⁻¹ which agree well with the structure 8.(c.f.Exp.)

The mass spectrum of 8 show the correct molecular ion peak at m/z = 456 (4.04 %) which upon loss of N- aminophthalimide radical afforded the base peak at m/z = 295 (100%).

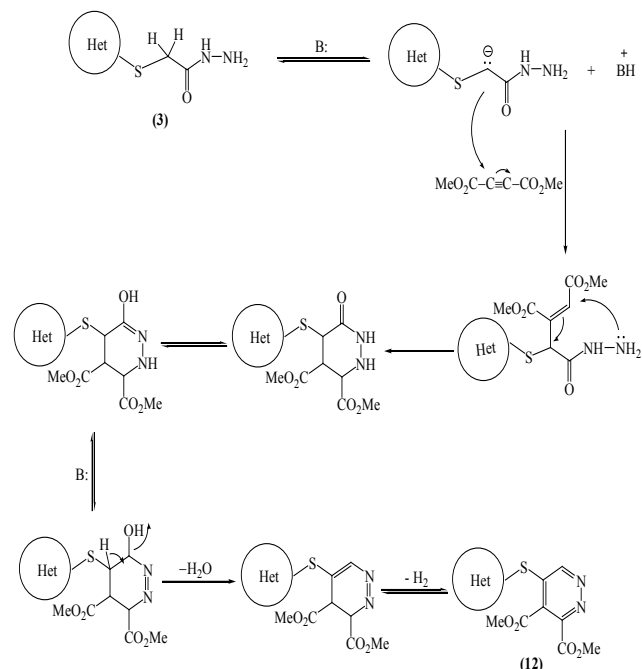
Refluxing compound 3 with acrylonitrile in dioxane afforded 2-[2-(5-iminopyrazolidin-1-yl)-2-oxoethyl thio]-3-phenylquinazolin-4(3H) one 10. The IR spectrum of 10 revealed the absence of the stretching absorption band for the nitrile group which indicates that the nitrile group involved in the cyclization process. The EI-MS is completely in accord with the assigned structure. (c.f. Exp.) Fusion of 3 with methylcinnamate on oil bath at 180 °C yielded a crude solid product which triturated with ethanol to give 2-[2-

oxo-2-(5-oxo-3-phenyl- Δ^3 pyrazolin-1-yl)ethyl thio]-3-phenylquinazolin-4(3H)one 11. The structure 11 was established by the spectroscopic and analytical data. $^1\text{H-NMR}$ spectrum of compound 11 (CDCl_3) lacked the signals attributable for ABX system $-\text{CH}-\text{CH}_2-$ which indicate the dehydrogenation during the reaction conditions. Ample evidence for the structure 11 is forthcoming from the analysis of the mass spectrum in which the highest recorded peak at $m/z = 455$ (64.9%) attributable for the $\text{M}+1$ radical cation. (c.f. Exp.) The reaction is smoothly proceeded via Michael addition followed by 1,5-exo-trig cyclization and dehydrogenation. (Scheme 5)



Scheme 5

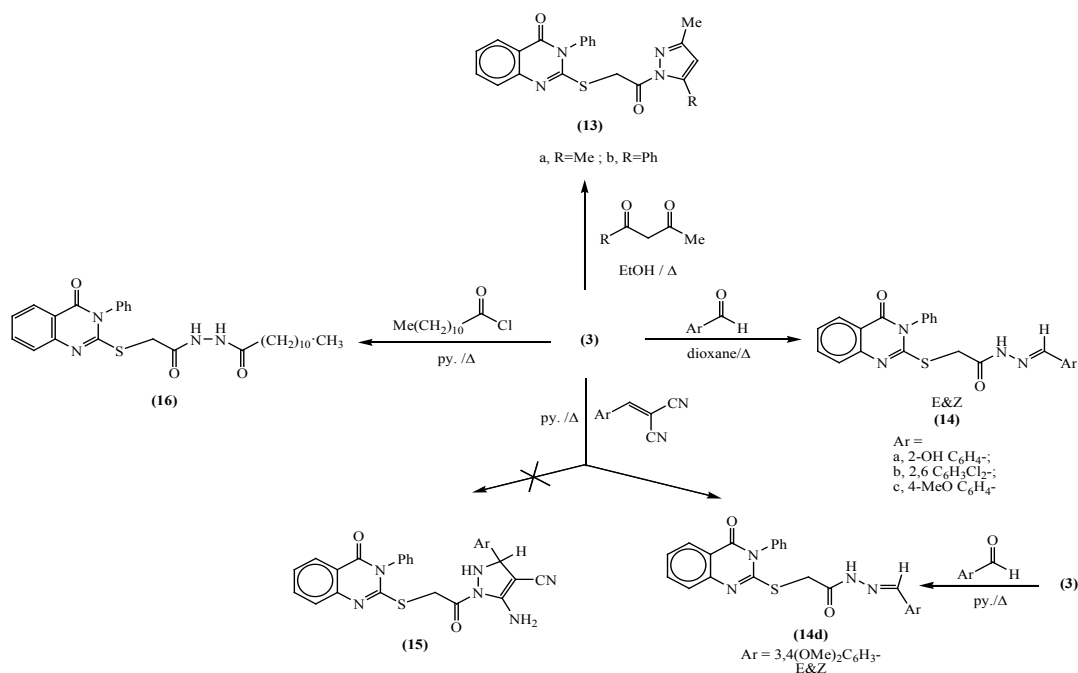
The reaction of 3 with dimethyl acetylene dicarboxylate (DMAD) gave a crude product with molecular formula $\text{C}_{22}\text{H}_{16}\text{N}_4\text{O}_5\text{S}$ (448) which indicate the removal of one molecule of water during the combination of the two substrates. This product was identified as diethyl-5-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio)pyridazin-3,4-dicarboxylate 12 using the analytical and spectroscopic data. (c.f. Exp.)



Scheme 6

The reaction is smoothly proceeded through Michael addition followed by cyclization. (Scheme 6)

Treatment of the thiohydrazone 3 with β -diketone such as acetyl acetone and/or benzoyl acetone in refluxing ethanol yielded 2-[2-(3,5-dimethylpyrazol-1-yl)-2-oxoethylthio]-3-phenyl-quinazolin-4(3H)-one 13a and 2-[2-(3-methyl-5-phenyl pyrazol-1-yl)-2-oxoethylthio]-3-phenyl-quinazolin-4(3H)-one 13b respectively. (Scheme 7) The structure 13a and 13b were established from analytical and spectroscopic data. The EI-MS of 13a and 13b have very close fragmentation pattern which were consistent with the assigned structure. (c.f. Exp.)



Scheme 7

The hydrazide 3 easily condensed with aromatic aldehydes such as salicylaldehyde, 2,6-dichlorobenzaldehyde and/or P-anisaldehyde in refluxing dioxane for 24hrs to give N'-Arylidene-2-(4-oxo-3-phenyl-3,4-dihydro-quinazolin-2-yl thio) acetohydrazide 14a-c. The structure 14 was established from analytical and spectroscopic data. Furthermore, the mass spectra of 14a-c show the correct molecular ion peaks and The EI-MS fragmentation pattern in all cases were very close as the base peak at $m/z = 295$ (100%). (c.f. Exp.)

Moreover, the $^1\text{H-NMR}$ spectra of compounds 14a and 14b showed two singlets for NH amide proton and two singlets for =CH olefinic proton, as well as, two singlets for -OH in case of 14a. This showed that compounds 14a and 14b were formed as a mixture of two geometrical (E- and Z-) isomers in which the E-isomers predominate.

The reaction of the hydrazide 3 with 3,4-dimethoxy α -cyano cinnamitrile in refluxing pyridine for 8 hrs followed by acidification with ice cold hydrochloric acid yielded the anil derivative 14d as a mixture of E- and Z-isomer and the pyrazolidine derivative 15 was not isolated. The IR spectrum of the product show the stretching absorption bands for ν_{NH} at 3180cm^{-1} , ν_{CO} at 1681cm^{-1} and lack the absorption band for ν_{NH_2} and $\nu_{\text{C=N}}$ which eliminate the assigned structure 15 and in accord with the proposed structure 14d.

$^1\text{H-NMR}$ spectrum is another clue for the structure 14d. (c.f. Exp)

Acylation of the hydrazide 3 with lauroyl chloride in refluxing pyridine yielded the N-acylated product 16.

The structure 16 was confirmed from the analytical and spectroscopic data. (c.f. Exp.)

3. Experimental

Melting points are measured on an electrothermal melting point apparatus. Elemental analyses were carried out at the microanalytical unit, Cairo Univeristy. The IR spectra were measured on a Unicam SP-1200 spectrometer using KBr Wafer technique. The $^1\text{H-NMR}$ spectra were measured in DMSO- d_6 on a Varian plus instrument (300MHz). Mass spectra were recorded on a shimadzu GC-MS QP- 1000EX instrument operating at 70 ev.

3.1. Synthesis of S-(Hetaryl)Thioglycollate 2

To a solution of 1 (2.54g, 0.01mol) and ethylchloroacetate (1.23g, 0.01mol) in (50ml) ethanol, fused sodium acetate (0.8g, 0.01mol) was added. The whole mixture was refluxed for 6hrs, the whole mixture was concentrated and left to cool. The precipitated solid was filtered off, washed with water, dried and recrystallized from ethanol to give ethyl-2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio)acetate 2.

Colorless crystals (70% yield); m.p: 105-107°C. IR: ν_{max} 1739 ($\text{C=O}_{\text{ester}}$), 1690 ($\text{C=O}_{\text{quinazolin}}$) cm^{-1} . $^1\text{H-NMR}$ (CDCl_3): δ (ppm) 1.31 (t, 3H, $\text{CH}_3\text{-CH}_2$, $J=7.2\text{Hz}$), 3.91 (s, 2H,

$\text{CH}_2\text{-CO}$), 4.24 (q, 2H, $\text{CH}_3\text{-CH}_2$, $J=7.2\text{Hz}$), 7.27-8.26 (m, 9H, ArH). MS: m/z (%) 340 (M^+ , 39), 295 (17), 294 (15), 267 (42), 266 (21), 253 (48), 165 (25), 221 (56), 119 (21), 90 (37), 77 (100), 63 (16). Anal. Calcd. for $\text{C}_{18}\text{H}_{16}\text{N}_2\text{O}_3\text{S}$ (340): C, 63.52; H, 4.70; N, 8.23; S, 9.41. Found C, 63.76; H, 4.95; N, 8.51; S, 9.79.

3.2. Hydrazinolysis of 2; Formation of Hydrazide 3

A mixture of 2 (2.5g, 0.008mol) and hydrazine hydrate (0.4g, 0.008mol) was stirred in (50ml) ethanol for 10hrs. The precipitated solid was filtered off and recrystallized from toluene to give 2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio)acetohydrazide 3.

Colorless crystals (65% yield); m.p: 180-183°C. IR: ν_{max} 3303, 3250, 3220 (NH, NH₂), 1693 ($\text{C=O}_{\text{quinazolin}}$), 1663 ($\text{C=O}_{\text{hydrazide}}$) cm^{-1} . $^1\text{H-NMR}$ (DMSO- d_6): δ (ppm) 3.83 (s, 2H, $\text{CH}_2\text{-CO}$), 4.25 (br.s, 2H, NH₂, exchangeable), 7.26-8.10 (m, 9H, ArH), 9.29 (br.s, 1H, NH-CO, exchangeable). MS: m/z (%) 326 (M^+ , 7), 296 (21), 295 (100), 254 (22), 253 (81), 221 (23), 132 (43), 90 (27), 77 (70). Anal. Calcd. for $\text{C}_{16}\text{H}_{14}\text{N}_4\text{O}_2\text{S}$ (326): C, 58.88; H, 4.32; N, 17.17; S, 9.82. Found C, 58.49; H, 4.10; N, 16.99; S, 9.60.

3.3. Reaction of the Hydrazide 3 with Carbon Disulphide in Pyridine; Formation of Oxadiazolthione Derivative 4

A solution of hydrazide 3 (0.65g, 0.002mol) in pyridine (50ml) and carbon disulphide (1.5ml, 0.02mol) were refluxed on water bath for 7hrs. The cooled concentrated solution was poured onto ice-water (50ml) and acidified with acetic acid (2ml). The mixture was extracted twice with chloroform (75ml). The organic layer was dried, concentrated, left for slow evaporation. The crude precipitated was crystallized from light petroleum to give 3-phenyl-2-((5-thio-4,5-dihydro-1,3,4-oxadiazol-2-yl)methylthio)quinazolin-4(3H)-one 4.

Yellow crystals (67% yield); m.p: 210-213°C. IR: ν_{max} 3444, 3130 (NH), 1665 ($\text{C=O}_{\text{quinazolin}}$), 1262 (C=S) cm^{-1} . $^1\text{H-NMR}$ (DMSO- d_6): δ (ppm) 4.53 (s, 2H, $\text{CH}_2\text{-S}$), 7.26-8.11 (m, 10H, 9ArH + NH exchangeable). MS: m/z (%) 368 (M^+ , 14), 266 (26), 264 (89), 253 (84), 230 (86), 221 (100), 167 (45), 144 (39), 143 (79), 119 (41), 91 (35), 77 (98), 63 (16). Anal. Calcd. for $\text{C}_{17}\text{H}_{12}\text{N}_4\text{O}_2\text{S}_2$ (368): C, 55.42; H, 3.28; N, 15.21; S, 17.41. Found C, 55.89; H, 3.41; N, 15.09; S, 17.67.

3.4. Reaction of Hydrazide 3 with Carbon Disulphide in Ethanolic Potassium Hydroxide; Formation of Quinazolidione Derivative 5

A solution of hydrazide 3 (0.65g, 0.002mol) in ethanol (50ml), potassium hydroxide (0.1g, 0.002mol) and excess of carbon disulphide (1.5ml, 0.02mol) were refluxed on water bath for 7hrs. The cooled concentrated solution was poured onto ice-water (50ml) and acidified with acetic acid (2ml). The crude product was filtered off, dried and recrystallized from ethanol to give 3-phenyl quinazolin-2,4(1H,

3H)-dione 5.

Yellow crystals (60% yield); m.p: 260-261°C[23]. IR: ν_{\max} 3368 (NH), 1731,1652 (C=O) cm^{-1} . MS: m/z (%)238 (M^+ ,100), 146 (42), 119(77), 92 (12). Anal. Calcd. for $\text{C}_{14}\text{H}_{10}\text{N}_2\text{O}_2$ (238): C, 70.58; H, 4.23; N, 11.76. Found C, 70.71; H, 4.40; N, 11.99.

3.5. Formation of Phenyl Isothio Semicarbazide Derivative 6

A mixture of hydrazide 3 (0.65g, 0.002mol) and phenyl isothiocyanate (0.2ml, 0.002mol) was refluxed in ethanol (30 ml) for 1hr. The reaction mixture was left to cool, the crude solid was filtered off, washed twice with cold ethanol, dried and recrystallized from ethanol to give 1-(2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio)acetyl)-4-phenyl thiosemicarbazide 6.

Yellow crystals (70% yield); m.p: 210-212°C. IR: ν_{\max} 3316, 3274, 3179 (NH), 1688 (C=O_{quinazolin}), 1657 (C=O_{amide}), 1617 (C=N), 1260 (C=S) cm^{-1} . MS: m/z (%) 461 (M^+ , 22), 420 (24), 367 (25), 312 (31), 295 (49), 294 (100), 293 (30), 260 (33), 253 (74), 239 (26), 180 (32), 150 (47), 134 (55), 94 (34). Anal. Calcd. for $\text{C}_{23}\text{H}_{19}\text{N}_5\text{O}_2\text{S}_2$ (461): C, 59.85; H, 4.15; N, 15.17; S, 13.89. Found C, 60.09; H, 4.31; N, 15.38; S, 13.57.

3.6. Cyclization of Compound 6; Formation of Triazolthione Derivative 7

A solution of 6 (0.7g, 0.0015mol) in ethanolic sodium hydroxide (0.25g in 20ml ethanol) was heated under reflux for 3hrs and then left to cool. The solid obtained was filtered off, washed with water and recrystallized from ethanol to give 3-phenyl-2-((4-phenyl-5-thioxo-4, 5-dihydro-1H-1, 2, 4-triazol-3-yl)methylthio)quinazolin-4(3H)-one 7.

Yellow crystals (79% yield); m.p >300°C. IR: ν_{\max} 3392 (NH), 1688 (C=O_{quinazolin}), 1247 (C=S) cm^{-1} . ¹H-NMR (DMSO-*d*₆): δ (ppm) 3.90 (s, 2H, CH₂-S-), 6.62-7.98 (m, 15H, 14ArH + NH exchangeable). MS: m/z (%)443 (M^+ , 14), 388 (21), 386 (100), 353 (24), 284 (41), 268 (40), 221 (69), 186 (46), 159 (43), 134 (72). Anal. Calcd. for $\text{C}_{23}\text{H}_{17}\text{N}_5\text{OS}_2$ (443): C, 62.28; H, 3.86; N, 15.79; S, 14.46. Found C, 62.17; H, 3.93; N, 15.90; S, 14.80.

3.7. Reaction of hydrazide 3 with Phthalic Anhydride; Formation of N-Substituted Isoindolindione Derivative 8

A mixture of hydrazide 3 (0.65g, 0.002mol) and phthalic anhydride (0.2g, 0.002mol) was refluxed in dry dioxane (20ml) for 5hrs. The reaction mixture was left overnight for slow evaporation. The crude product was recrystallized twice from light petroleum /toluene mixture to give N-(1,3-dioxoisindolin-2-yl)-2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio)acetamide 8. Yellowish white crystals (50% yield); m.p: 250-252°C. IR: ν_{\max} 3185 (NH), 1793, 1738, 1684 (C=O) cm^{-1} . ¹H-NMR (DMSO-*d*₆): δ (ppm) 3.99 (s, 2H, CH₂-S-), 7.27-7.89 (m, 13H, ArH), 9.82 (s, 1H, NH, exchangeable). MS: m/z (%) 456 (M^+ , 4), 296 (27), 295

(100), 253 (17), 221 (21), 132 (39), 104 (48), 89 (20). Anal. Calcd. for $\text{C}_{24}\text{H}_{16}\text{N}_4\text{O}_4\text{S}$ (456): C, 63.15; H, 3.53; N, 12.27; S, 7.02. Found C, 63.27; H, 3.68; N, 12.08; S, 6.90.

3.8. Synthesis of Iminopyrazolidin Derivative 10

A mixture of hydrazide 3 (0.65g, 0.002mol), acrylonitrile (0.114g, 0.002mol) in pyridine

(20ml) was refluxed for 3hrs. The reaction mixture was left overnight for slow evaporation. The formed solid residue was recrystallized from light petroleum-toluene mixture to give 2-(2-(5-iminopyrazolidin-1-yl)-2-oxoethyl thio)-3-phenyl quinazolin-4(3H)-one 10.

Yellow crystals (70% yield); m.p: 224-226°C. IR: ν_{\max} 3190 (NH), 1687 (C=O) cm^{-1} . MS: m/z (%)379 (M^+ , 14), 295 (31), 254 (49), 253 (90), 235 (30), 225 (16), 221(49), 181 (21), 167 (90), 166 (50), 162 (36), 145 (31), 144 (100), 134 (20), 132 (28), 119 (45), 92 (40), 90 (59), 77 (75). Anal. Calcd. for $\text{C}_{19}\text{H}_{17}\text{N}_5\text{O}_2\text{S}$ (379): C, 60.14; H, 4.52; N, 18.46; S, 8.45. Found C, 60.39; H, 4.87; N, 18.79; S, 8.20.

3.9. Reaction of 3 with Methyl Cinnamate; Formation of Pyrazolidinone Derivative 11

A mixture of hydrazide 3 (0.65g, 0.002mol) and methyl cinnamate (0.23g, 0.002mol) was fused in sand bath for 4hrs at 180-190°C. The residue was treated with hot light petroleum/toluene mixture (50ml). The formed crystals were collected, dried and crystallized from toluene to give 2-(2-oxo-2-(5-oxo-3-phenyl-2H-pyrazol-1(5H)-yl) ethyl thio)-3-phenyl quinazolin-4(3H)-one 11.

Yellow crystals (56% yield); m.p: 258-260°C. IR: ν_{\max} 2220 (NH), 1690, 1641 (C=O) cm^{-1} . ¹H-NMR (DMSO-*d*₆): δ (ppm) 3.78 (s, 2H, CH₂-S-), 7.26-8.23 (m, 15H, ArH), 10.71 (br.s, 2H, NH, exchangeable). MS: m/z (%)454 (M^+ , 65), 253 (93), 231 (76), 221 (92), 145 (67), 143 (79), 134(68), 109 (67), 105 (67), 85 (67), 77 (83), 75 (91), 62 (75), 61 (100). Anal. Calcd. for $\text{C}_{25}\text{H}_{18}\text{N}_4\text{O}_3\text{S}$ (454): C, 66.07; H, 3.99; N, 12.33; S, 7.05. Found C, 66.29; H, 4.15; N, 12.70; S, 7.19.

3.10. Synthesis of Pyridazine Dicarboxylate Derivative 12

A solution of hydrazide 3 (0.65g, 0.002mol)and (0.22g, 0.002mol) dimethyl acetylene dicarboxylate in dioxane (20ml) was refluxed for 6hrs. The reaction mixture was left overnight for slow evaporation. The crude precipitate was crystallized from benzene to give dimethyl-5-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-yl thio)-pyridazin-3,4- dicarboxylate 12.

Yellowish white crystals (60% yield); m.p: 127-129°C. IR: ν_{\max} 3437, 3261 (NH), 1734, 1689 (C=O) cm^{-1} . ¹H-NMR (DMSO-*d*₆): δ (ppm) 3.64 (s, 3H, -COOCH₃) , 3.78 (s, 3H, -COOCH₃), 7.08-7.81 (m, 9H, ArH), 8.07 (s, 1H, C₆-H_{pyridazin}). MS: m/z (%)448 (M^+ , 20), 435 (17), 423 (15), 409 (80), 394 (17), 379 (21), 362(19), 295 (25), 253 (16), 295 (65), 267 (14), 254 (28), 253 (59), 235 (16), 221 (41), 191 (34), 181 (43), 119 (23), 111 (37), 110 (44), 91

(50), 77 (100). Anal. Calcd. for $C_{22}H_{16}N_4O_5S$ (448): C, 58.92; H, 3.60; N, 12.49; S, 7.15. Found C, 59.17; H, 3.81; N, 12.69; S, 7.40.

3.11. General Procedure for the Reaction of Hydrazone 3 with Acetyl Acetone or Benzoyl Acetone; Formation of Pyrazol Derivative 13

A mixture of hydrazone 3 (0.65g, 0.002mol), acetyl acetone (0.2ml, 0.002mol) or benzoyl acetone (0.3g, 0.002mol) was refluxed in ethanol (50ml) for 7hrs. The reaction mixture was left overnight for slow evaporation. The solid product was collected, filtered off, dried and recrystallized from light petroleum and chloroform to give 13a and 13b respectively.

3.11.1. 2-(2-(3,5-dimethyl-1H-pyrazol-1-yl)-2-oxo ethylthio)-3-phenyl quinazolin-4(3H)-one 13a

Pale Yellow crystals (78% yield); m.p: 76-78°C. IR: ν_{max} 1688 ($C=O_{quinazolin}$), 1644 ($C=O_{amide}$) cm^{-1} . ^1H-NMR ($CDCl_3$): δ (ppm) 1.82 (s, 3H, C_5-Me), 2.08 (s, 3H, C_3-Me), 4.13 (s, 2H, CH_2-S) 7.2-7.6 (m, 9H, ArH), 8.2 (s, 1H, $C_4-H_{pyrazole}$). MS: m/z (%) 390 (M^+ , 8), 296 (27), 295 (100), 294 (33), 255 (13), 253 (24), 221 (24), 132 (12). Anal. Calcd. for $C_{21}H_{18}N_4O_2S$ (390): C, 64.60; H, 4.65; N, 14.35; S, 8.25. Found C, 64.97; H, 4.37; N, 14.72; S, 8.53.

3.11.2. 2-(2-(3-methyl-5-phenyl-1H-pyrazol-1-yl)-2-oxo ethylthio)-3-phenyl quinazolin-4(3H)-one 13b

Bright Yellow crystals (55% yield); m.p: 166-168°C. IR: ν_{max} 1689 ($C=O_{quinazolin}$), 1652 ($C=O_{amide}$) cm^{-1} . MS: m/z (%) 452 (M^+ , 10), 439 (20), 411 (19), 403 (23), 340 (46), 328 (40), 307 (52), 294 (42), 253 (100), 235 (65), 221 (66), 162 (64), 149 (80), 147 (66), 132 (47), 104 (49), 97 (53). Anal. Calcd. for $C_{26}H_{20}N_4O_2S$ (452): C, 69.01; H, 4.45; N, 12.38; S, 7.09. Found C, 69.30; H, 4.72; N, 12.19; S, 6.95.

3.12. General Procedure for the Reaction of Hydrazone 3 with Aromatic Aldehydes

A solution of hydrazone 3 (0.65g, 0.002mol) in dioxane (30 ml) and salicylaldehyde (0.2g, 0.002mol) or 2,6-dichlorobenzaldehyde (0.3g, 0.002mol) or p-anisaldehyde (0.12g, 0.002mol) was refluxed in presence of (0.2ml) triethylamine for 24h. The precipitate formed was dried and recrystallized from light petroleum-toluene mixture to give N' -arylidene-2-(4-oxo-3-phenyl-3, 4-dihydroquinazolin-2-ylthio)acetohydrazone 14a, 14b and 14c respectively.

3.12.1. N' -(2-hydroxybenzylidene)-2-(4-oxo-3-phenyl-3, 4-dihydroquinazolin-2-ylthio) acetohydrazone 14a

Yellow crystals (85% yield); m.p: 215-217°C. IR: ν_{max} 3444 (OH), 3177 (NH), 1678 ($C=O_{quinazolin}$), 1655 ($C=O_{amide}$) cm^{-1} . ^1H-NMR ($DMSO-d_6$): δ (ppm) (E-form, 55%) 4 (s, 2H, $-CH_2-S$), 6.89-8.09 (m, 13H, ArH), 8.47 (s, 1H, =CH), 10.99 (s, 1H, -OH, exchangeable), 11.95 (s, 1H, NH, exchangeable); (Z-form, 45%) 8.34 (s, 1H, =CH), 10.08 (s, 1H, -OH, exchangeable), 11.60 (s, 1H, NH, exchangeable). MS:

m/z (%) 430 (M^+ , 8), 297 (10), 296 (21), 295 (100), 253 (30), 132 (40), 77 (48). Anal. Calcd. for $C_{23}H_{18}N_4O_3S$ (430): C, 64.17; H, 4.21; N, 13.01; S, 7.45. Found C, 64.40; H, 4.45; N, 12.89; S, 7.67.

3.12.2. N' -(2,6-dichlorobenzylidene)-2-(4-oxo-3-phenyl-3, 4-dihydroquinazolin-2-ylthio) acetohydrazone 14b

Yellow crystals (80% yield); m.p: 208-210°C. IR: ν_{max} 3189 (NH), 1684 ($C=O_{quinazolin}$), 1657 ($C=O_{amide}$) cm^{-1} . ^1H-NMR ($DMSO-d_6$): δ (ppm) (E-form, 82%) 4.43 (s, 2H, $-CH_2-S$), 7.43-8.08 (m, 12H, ArH), 8.31 (s, 1H, =CH), 11.87 (s, 1H, NH, exchangeable); (Z-form, 18%) 8.45 (s, 1H, =CH), 12.02 (s, 1H, NH, exchangeable). MS: m/z (%) 482 (M^+ , 0.86), 451 (15), 409 (10), 311 (17), 297 (43), 295 (100), 268 (18), 264 (15), 221 (17), 132 (43), 91 (18), 90 (22), 77 (47). Anal. Calcd. for $C_{23}H_{16}Cl_2N_4O_3S$ (482): C, 57.15; H, 3.34; N, 11.59; S, 6.63. Found C, 57.40; H, 3.59; N, 11.77; S, 6.39.

3.12.3. (E)- N' -(4-methoxybenzylidene)-2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-ylthio) acetohydrazone 14c

Yellow crystals (82% yield); m.p: 217-219°C. IR: ν_{max} 3172 (NH), 1680 ($C=O_{quinazolin}$), 1645 ($C=O_{amide}$) cm^{-1} . MS: m/z (%) 444 (M^+ , 20), 296 (20), 295 (100), 253 (25), 221 (16), 144 (29), 132 (26), 120 (23), 104 (18), 91 (30), 77 (38). Anal. Calcd. for $C_{24}H_{20}N_4O_3S$ (444): C, 64.85; H, 4.54; N, 12.60; S, 7.21. Found C, 65.19; H, 4.80; N, 12.93; S, 7.09.

3.13. The Reaction of Hydrazone 3 with 2-(3, 4-dimethoxybenzylidene)Malononitrile. Formation of 14d

A solution of hydrazone 3 (0.65g, 0.002mol) and 2-(3,4-dimethoxy benzylidene)malononitrile (0.2g, 0.002 mol) in pyridine (10ml) was refluxed for 8hrs. The reaction mixture was acidified with ice cold hydrochloric acid. The solid product obtained was filtered off and recrystallized from ethanol to give 14d as a mixture of E&Z isomers. The same product 14d was obtained when the hydrazone 3 refluxed with 3,4-dimethoxy benzaldehyde in pyridine.

3.13.1. N' -(3,4-dimethoxybenzylidene)-2-(4-oxo-3-phenyl-3,4-dihydroquinazolin-2-ylthio) acetohydrazone 14d

Yellow crystals (50% yield); m.p: 220-222°C. IR: ν_{max} 3180 (NH), 1681 ($C=O_{quinazolin}$), 1645 ($C=O_{amide}$) cm^{-1} . ^1H-NMR ($CDCl_3$): δ (ppm) (E-form, 75%) 3.88 (s, 3H, OCH_3), 3.94 (s, 3H, OCH_3), 4.5 (s, 2H, $-CH_2-S$), 6.81-8.34 (m, 12H, ArH), 8.93 (s, 1H, =CH), 10.79 (s, 1H, NH, exchangeable); (Z-form, 25%) 3.79 (s, 3H, OCH_3), 3.91 (s, 3H, OCH_3), 8.89 (s, 1H, =CH), 10.75 (s, 1H, NH, exchangeable). Anal. Calcd. for $C_{25}H_{22}N_4O_4S$ (474): C, 63.28; H, 4.67; N, 11.81; S, 6.76. Found C, 63.51; H, 4.92; N, 11.59; S, 6.55.

3.14. Acylation of 3 Using Lauroyl Chloride

A mixture of hydrazone 3 (0.33 g, 0.001 mol) in dry pyridine (20 ml) and lauroyl chloride (0.22 g, 0.001 mol) were

refluxed for 4h. The reaction mixture poured onto ice cold acetic acid. The solid product was filtered off and recrystallized from petroleum ether/toluene mixture to give N-acylated product 16.

3.14.1. N'-(2-(4-oxo-3-phenyl-3, 4-dihydroquinazolin-2-ylthio)acetyl)dodecane hydrazide 16

Yellowish white crystals (65% yield); m.p: 120-121°C. IR: ν_{\max} 3210, 3195 (NH), 1701 (C=O_{quinazolin}), 1650 (C=O_{amide}) cm^{-1} . $^1\text{H-NMR}$ (DMSO- d_6): δ (ppm) 0.87 (t, 3H, CH₃, J = 6.6Hz), 1.24-1.48 (m, 18H, 9 CH₂), 2.05-2.17 (m, 2H, CH₂CO), 3.94 (s, 2H, CH₂-S), 7.44-8.01 (m, 9H, ArH), 9.80 (s, 1H, NH, exchangeable), 10.1 (s, 1H, NH exchangeable). Anal. Calcd. for C₂₈H₃₆N₄O₃S (508): C, 66.11; H, 7.13; N, 11.01; S, 6.30. Found C, 66.40; H, 7.39; N, 11.21; S, 6.17.

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