Production of Xanthan employing \textit{Xanthomonas campestris} using Sugarcane Molasses

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\textbf{Abstract} \ Xanthan production by \textit{Xanthomonas campestris} from pre-treated sugarcane molasses (acidified molasses and acidified aerated molasses) was investigated. The optimization of xanthan yield was done at different pH, temperature, and incubation time for both the pre-treated sugarcane molasses. Maximum yield was achieved in 1\% acidified molasses and acidified aerated molasses with 1\% yeast extract as nitrogen source for 3 days of incubation time at 30°C. The precipitation of xanthan gum was done after 48 hours and the yield was high in acidified molasses and acidified aerated molasses comparing to acidified sugarcane molasses. \textit{Xanthomonas campestris} produced xanthan yield of 12.23 g/l using acidified molasses and acidified aerated molasses. This fermentation study has an advantage over some other manufacturing processes with its use of agro industrial wastes as the raw material, allowing the increased xanthan production.

\textbf{Keywords} \ \textit{Xanthomonas campestris}, sugarcane molasses, xanthan, acidified molasses, acidified aerated molasses

1. Introduction

Xanthan gum is the most commercially accepted microbial extracellular heteropolysaccharide used in oil, food, and pharmaceutical industries as viscosifier, thickener, stabilizer, and emulsifier [1, 2]. The efficient xanthan gum producer is \textit{Xanthomonas campestris}. Other species such as \textit{X. phaseolii}, \textit{X. malvacearum}, \textit{X. carotae}, \textit{X. manihotis} and \textit{X. juglandis} are known to produce this exopolysaccharide as well. At present the annual production capacity is 5000 tons in India. The demand for xanthan is increasing every year. Research on the biosynthesis process has been recently focused on the biochemistry of the production[3] and the development of several culture methods to improve yields[4], and the search of \textit{X. campestris} mutants with increased xanthan production [5]. Most studies report the clear dependency between strain used and its xanthan yield and properties[6].

The nutritional veracity of \textit{X. campestris} allows the use of industrial, complex, and synthetic media formulations. The industrial scale production of xanthan is carried out using inexpensive substrates and nutrients. Efficient conversion of carbon source to the desired polysaccharide production requires a high carbon to nitrogen ratio. Sugarcane molasses is the main by product in sugarcane industry having a rich source of nutrients that lead environmental pollution.
nitrogen, sulphides, oils and grease were analyzed following the standard methods[7].

2.3. Pre-treatment of Sugarcane Molasses

Two types of pre-treatment methods were adopted for sugarcane molasses. 200 ml of the sugarcane molasses was acidified with 10 ml of 5 N H$_2$SO$_4$ and it was precipitated at pH 4 and stored (Acidified sugarcane molasses – ASM) and ASM was then aerated by air sparing, 3 ppm for 15 min at 80°C and stored (Acidified, aerated sugarcane molasses – AASM).

2.4. Optimization of Sugarcane Molasses and Nitrogen Source Concentration

The optimum concentration of ASM and AASM as sole carbon source and yeast extract as nitrogen source were determined by culturing 5% (v/v) of inoculum in medium (K$_2$HPO$_4$-2g/l, MgSO$_4$-0.1g/l, Tween 80-300 ppm) amended with different concentrations (0.5%, 1%, 1.5% and 2%) of ASM, AASM, and yeast extract, incubated at 30°C for 48 h. Cell growth was monitored by measuring the optical density (OD) at 420 nm (OD420; UV – Visible spectrophotometer-Genesys 2, Spectronic Instruments, NY) at 24 h and 48 h. Dry cell weight (DCW) was also estimated at 48 h[8]. The xanthan produced was recovered from cell-free supernatant by precipitation with two volumes of ice cold isopropanol with addition of 1% (w/v) KCl and dry weight was obtained by freeze-drying of xanthan precipitated according to the methods of Lopez et al.[9].

2.5. Optimization of pH, Temperature and Incubation Period

Medium for optimization studies (ASM/AASM-10g/l, yeast extract-10g/l, K$_2$HPO$_4$-2g/l, MgSO$_4$-0.1g/l, Tween–80 ppm) was prepared with different pH 6.6, 6.8, 7.0, and 7.2, and 5% (v/v) of 48 h grown culture was inoculated and incubated at 30±2°C. For optimization of temperature, the medium with 5% (v/v) inoculum was incubated at different temperatures such as 26°C, 28°C, 30°C and 32°C for 48 h. Xanthan yield, OD and DCW were estimated after 24 and 48 h following the standard procedures. For optimization of incubation period, OD, DCW and xanthan yield under the optimized conditions were estimated at 36, 40, 44, 48 and 52 h.

3. Results and Discussion

The characteristics of sugarcane molasses taken as the carbon source were analysed and the results were noted. The optimized concentration of ASM and AASM as carbon source was found to be 1% which showed maximum cell biomass and xanthan yield (Figure 1). Various concentrations of yeast extract influenced the growth of organism and xanthan yield (Figure 2). The xanthan yield was high at 1% yeast extract. The best suited temperature for bacterial growth and xanthan production was found to be 30°C followed by room temperature, 25°C and 45°C (Figure 3). The best bacterial growth and xanthan production followed a pH 9> 6.2 > 6.4 > 6.6 > 7 > 7.2 sequence (Figure 4). The xanthan production increased from 24 h to 48 h of growth and declined thereafter. X. campestris produced high xanthan yield of 12.23 and 10.3g/l with 1% AASM and ASM broth respectively under optimized condition.
The parameters such as carbon source, nitrogen source, pH, temperature and incubation time influenced the production rate of xanthan. Dissociation of cell growth and metabolite production allow the creation of continuous reactors where conversion yields of carbon to product were very high. This was especially true for products such as xanthan, where physical parameters change rapidly especially viscosity[10]. Moreover production was promoted at 30°C while 25°C was the optimum temperature for cell growth [11]. Similar condition was also noticed in the present study where 30°C was found to be the optimized temperature for the maximum production of xanthan at 48 h of incubation time. Mineral sources such as CaCO₃ and KH₂PO₄ proved to be important factors influencing the polysaccharide production and quality. However if the complex nutrients were added in lower quantities, a slight stimulation of xanthan formation was obtained [11]. A two-stage batch fermentation of X. campestris in a Glucose/Yeast extract shift from an initial low level (2.5% glucose/0.3% yeast extract) to a high level (5.0%glucose/0.3%yeast extract) at the end of the exponential growth phase was found to be preferable for xanthan production[12].

The maximum amount of cell concentration and xanthan formation occurs at 30g/l of glucose concentration and when glucose concentration is increased further[13]. It is concluded that the strain X. campestris produced maximum xanthan in high glucose concentrations at 30-40g/kg broth [14]. Xanthan production was maximum at 96 h using X. campestris [15]. Maximum cell concentration (1.5g/l) and xanthan yield (3.0g/l) was observed respectively when grown in 0.5 to 4.50g/l of yeast extract [16]. Figure 3 illustrates the production of xanthan at different pH and temperature. However, the optimum xanthan yield (13.0g/l) was obtained at pH (7.0) and temperature (30°C). Xanthan production by X. campestris was maximum at 28-30°C, whereas the cell growth was slow at 35°C[15]. West[16] reported that the optimal temperature for X. campestris growth was 28°C, whereas the maximum xanthan production was between 30°C and 33°C. Polysaccharide production by the Sphingo-

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**REFERENCES**


